

Water Desalination

Energy & Environmental Challenges



Methods of desalination



Thermal methods:

MED/MSF

Feed>Heat > Vapor > Fresh water

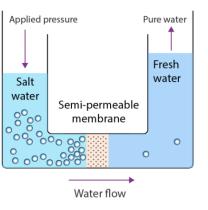
Vacuum freezing (VFVC)

Feed>Freezing>Vacuum>Fresh water



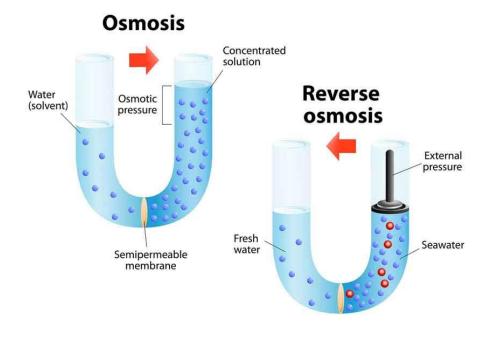
Reverse Osmosis method:

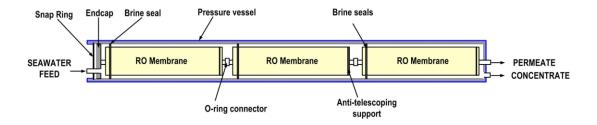
Feed>Pressure>Fresh water



How does reverse osmosis work?

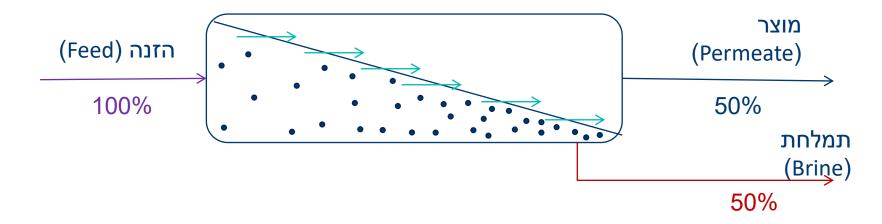






Recovery Rate





$$\% Recovery = \frac{Q_{Permeate}}{Q_{Feed}} = \frac{Q_{Permeate}}{Q_{Permeate} + Q_{Brine}}$$

Energy Challenges



- Fact:
- Large desalination facilities desalinate ~50,000 m^3/h
- ± Olympic pool every 2 minutes...

- Problems:
- How much energy it takes to evaporate that much water? (~1.15 US\$ per m^3)
- How much energy it takes to pressurize that much water? (~0.63 US\$ per m^3)
- Specific energy = ~3.2 kwh/m^3

Environmental Challenges



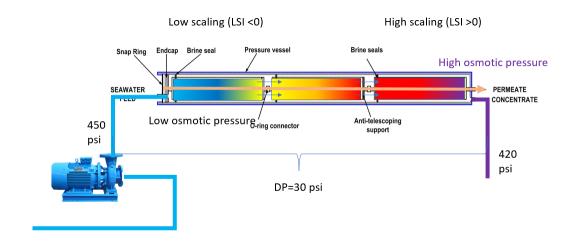
- Fact:
- Large desalination facilities desalinate ~50,000 m^3/h
- ± Olympic pool every 2 minutes...
- 50% of the water will go back as brine

- Problems:
- Chemicals as antiscaling agents (Phosphates), flocculants (Fe2SO4) and acids must be added.

Why won't we work in higher recovery?



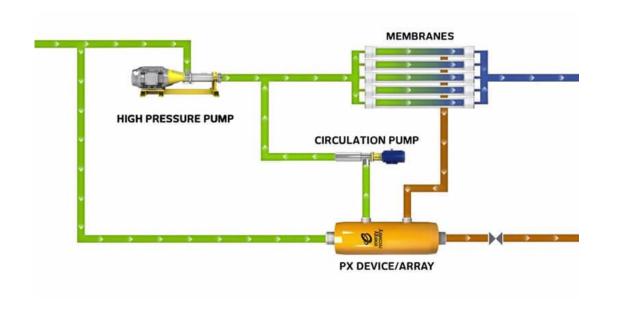
- Membrane scaling
- Higher concentration of brine stream
- Higher energy consumption



Energy Recovery

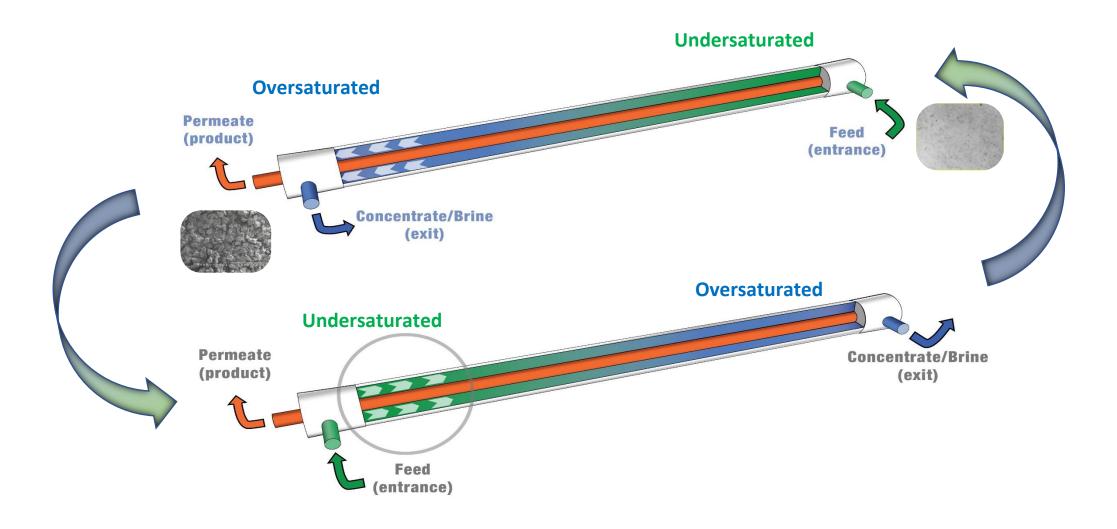


- Using the pressure from brine side to help pressurize inlet stream.
- 1) High pressure brine IN
- 2) Low pressure feed IN
- 3) High pressure feed OUT
- 4) Low pressure brine OUT



Extreme High Recovery without Scaling





ZLD / MLD

OROTECby WFI Group

- Extreme high recovery can be used before evaporators / crystallizers and filter press in order to achieve Zero Liquid Discharge
- Still today require very high CAPEX and even higher OPEX

